



Vth International summer school of  
Environmental Sciences (ISU Eco 2013)

# Energy storage with supercapacitors

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Martin-Luther-Universität  
Halle-Wittenberg



Interdisciplinary Center  
of Materials Science

# Center of Materials Science (CMAT)

**Nanotechnikum Weinberg**, since 2008, 1800 m<sup>2</sup> labs, 620 m<sup>2</sup> cleanroom



## Central lab units (CMAT-MLU Halle)

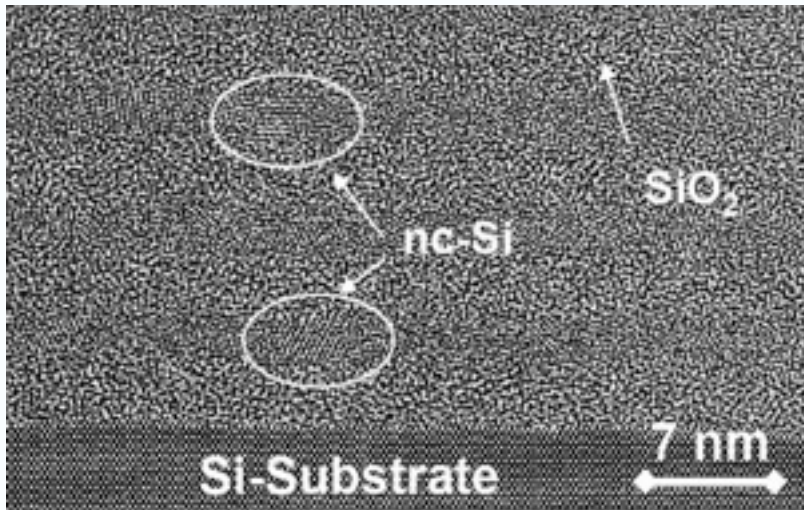
- ◆ Nanostructuring: lithography, thin film deposition, device prototyping
- ◆ Nanoanalysis: electron microscopy, optical characterization, positron annihilation

## Research disposal area (Bio-Nano Center)

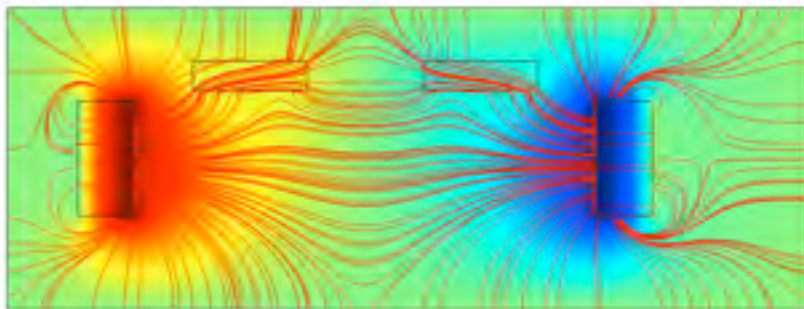
for physicists, chemists, materials scientists, biologists, pharmacists  
MLU, MPI, Fraunhofer, TGZ (KMU)



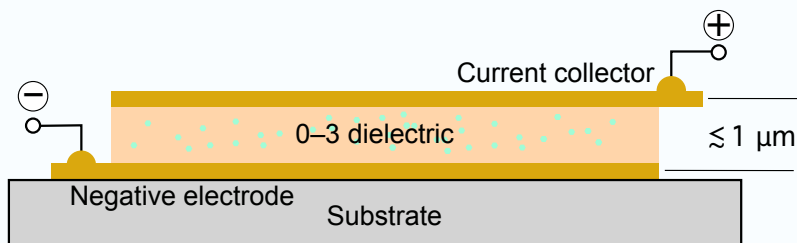
# Renewable energy materials



- ◆ Nanostructured thin film materials as functional elements for next-generation solar cells



- ◆ Silicon-related materials for thermoelectric applications



- ◆ Novel supercapacitors as energy storage devices



GEFÖRDERT VOM



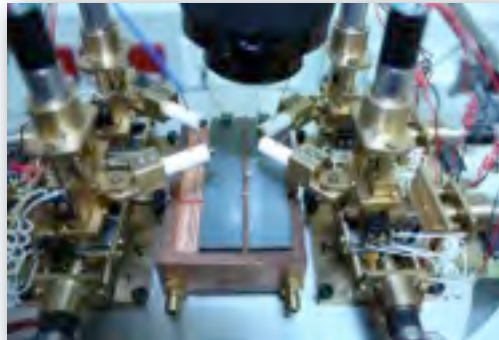
# Equipment

## Nanostructuring



Cleanroom class 10/100/10000

## Analysis



- ◆ Various electron microscopes
- ◆ Raman microscopy, ellipsometry
- ◆ Atomic force microscopy
- ◆ Electrical/thermal transport measurements

# Nanostructured materials

- ◆ Better energy storage devices are needed for sustainable energy supply.
- ◆ New materials are the key for basic improvements.
- ◆ **Nanoscaled materials** can be precisely adopted for energy harvesting, transformation and storage.
- ◆ Excellent properties for the selection of electrodes, electrolytes or dielectrics
- ◆ Nano-scaled electrolytes, nanoelectrodes for lithium ion batteries, supercapacitors, fuel cells
- ◆ The same concept is followed for electrochemical, as well as for electrostatic storage devices.

# Renewable energies = Direct energy from the sun

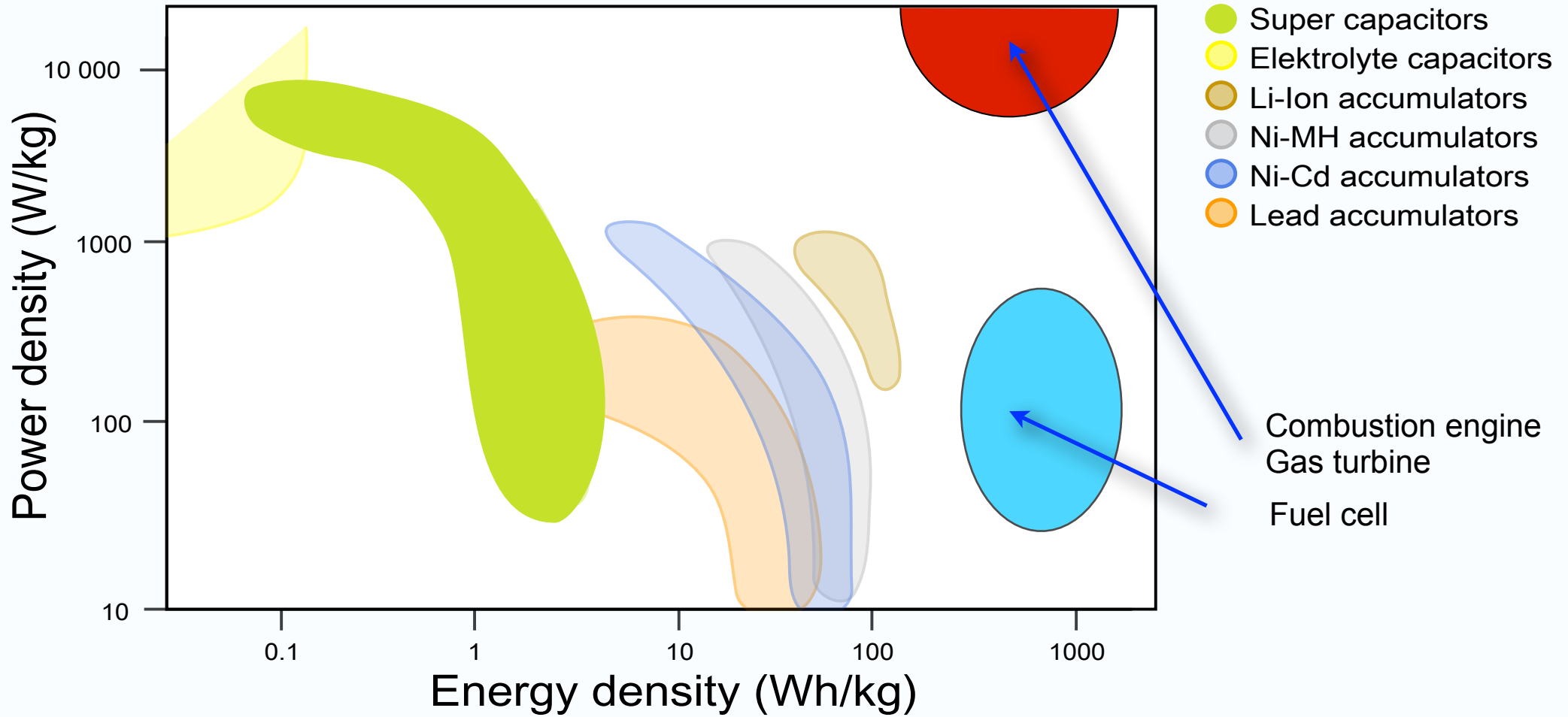
- ◆ Oil resources: 3 trillion barrels ( $4 \cdot 10^{14}$  kg)  $\cong$  energy of  $2 \cdot 10^{22}$  J;  
supplied from the Sun in 1½ days
- ◆ Amount of energy humans use annually:  $5 \cdot 10^{20}$  J,  
delivered to Earth by the Sun in 1 h
- ◆ Enormous power of the Sun continuously delivered to Earth:  
 $1 \cdot 10^5$  TW; human civilization uses currently 10 TW

# Energy storage

- ◆ Renewable energy sources: highly discontinuous
- ◆ Various energy storage concepts
  - ❖ Thermal and thermochemical storage (water, water–gravel, latent heat)
  - ❖ Chemical storage (hydrogen)
  - ❖ Mechanical storage (fly wheel, pump storage station, compressed air)
  - ❖ Electrochemical storage (lead, lithium ion, redox flow, NaS battery)
- ◆ Advantages ↔ disadvantages
  - no single solution for all applications



# Ragone diagram





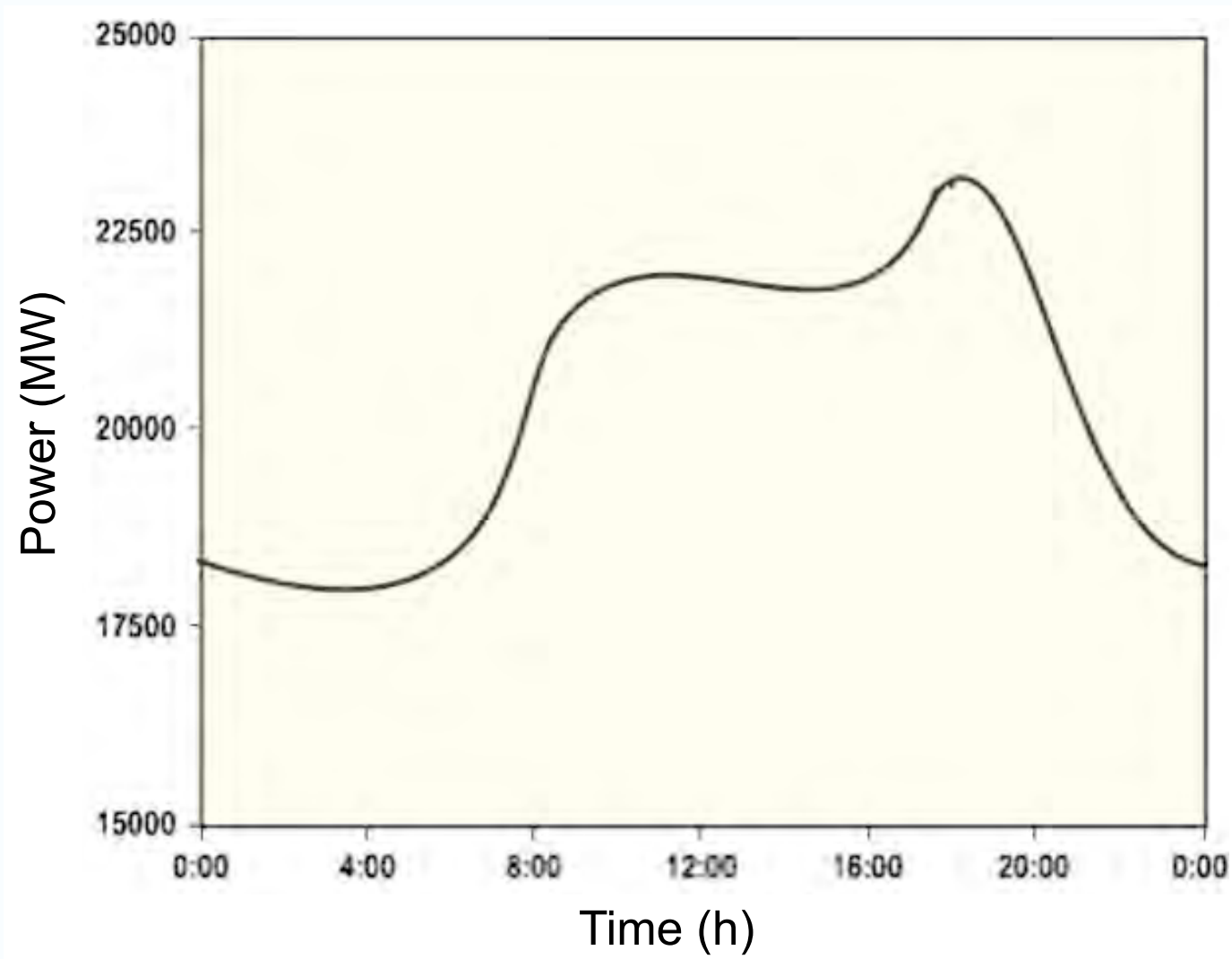
# Time scales



- ◆ Large time scales (seconds to weeks)
- ◆ Short-time storage  
(fluctuations in the grid, grid management, guarantee of supply)
- ◆ Middle-range storage  
(electromobility)
- ◆ Long-time storage  
(e. g. longer periods without wind)



# Need for energy storage



Example of the time dependence of the daily electrical power demand

[Huggins 2010]

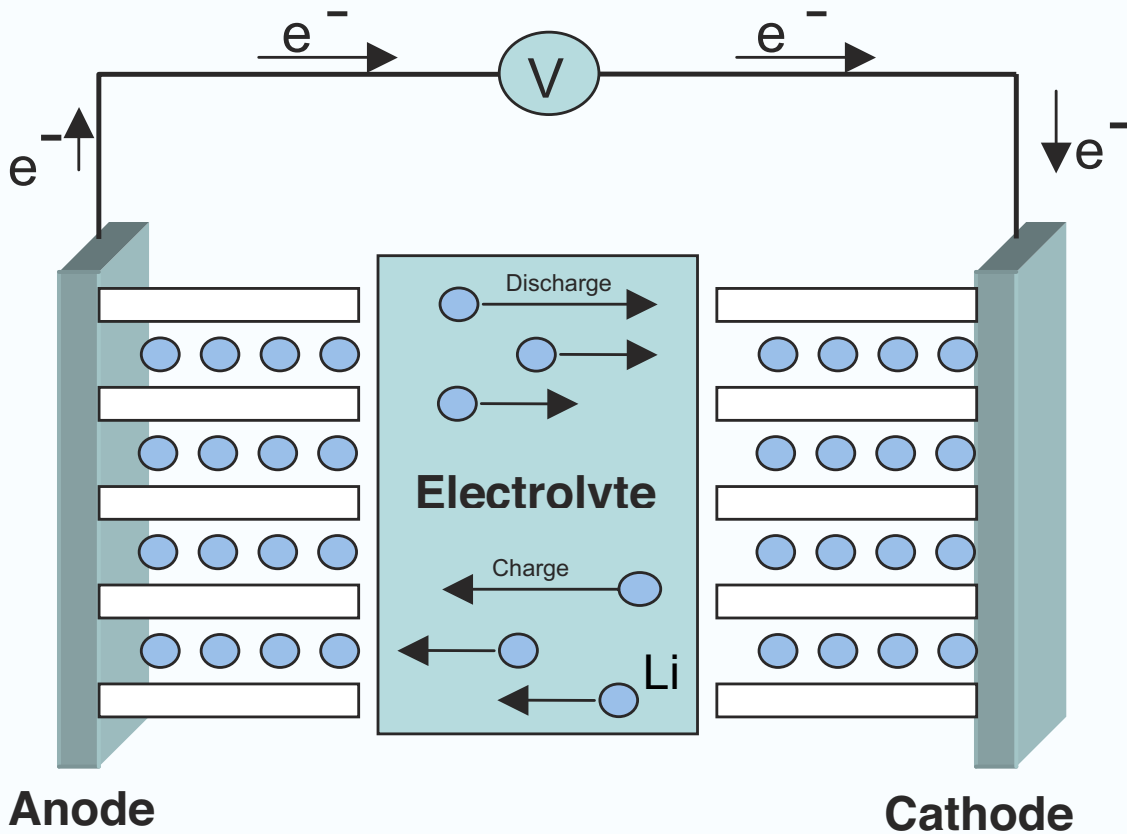
# Electrical storage

- ◆ Characteristics
  - ❖ Energy density, power density, storage time, voltage
  - ❖ Industrial processing, prize, weight
- ◆ Electrochemical devices (batteries, accumulators) mainly used
- ◆ Disadvantages
  - ❖ Limited lifetime, temperature range
  - ❖ Memory effect
  - ❖ Problems with overloading, deep discharge
  - ❖ Low charging speeds

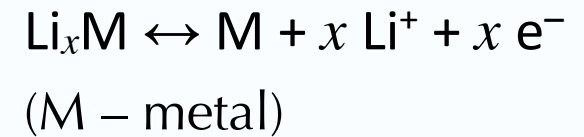


Selfdischarge	Battery:	1 – 5 %	per year
	Accum:	Li Ion: 2 %	per month
		Lead: 2 – 30 %	
		NiCd: 15 – 20 %	

# Lithium ion battery



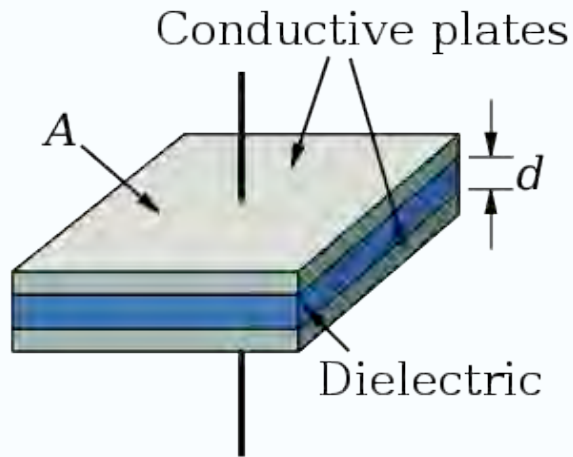
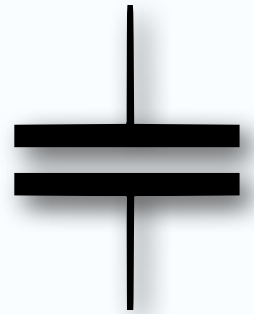
## Classical electrode process (Intercalation)



Scheme of a classical LIB  
[Wallace 2009]

# Capacitors

Capacitance  $C$  = Amount of charge stored per unit voltage



$$C = \epsilon_r \epsilon_0 \frac{A}{d}$$

$\epsilon_0$  dielectric constant  $\approx 9 \cdot 10^{-12}$  F/m

$\epsilon_r$  relative static permittivity of the dielectric  
(sometimes called dielectric constant)

Energy stored:

$$E = \frac{1}{2} C U^2 = \frac{1}{2} \epsilon_r \epsilon_0 \frac{A}{d} U^2$$

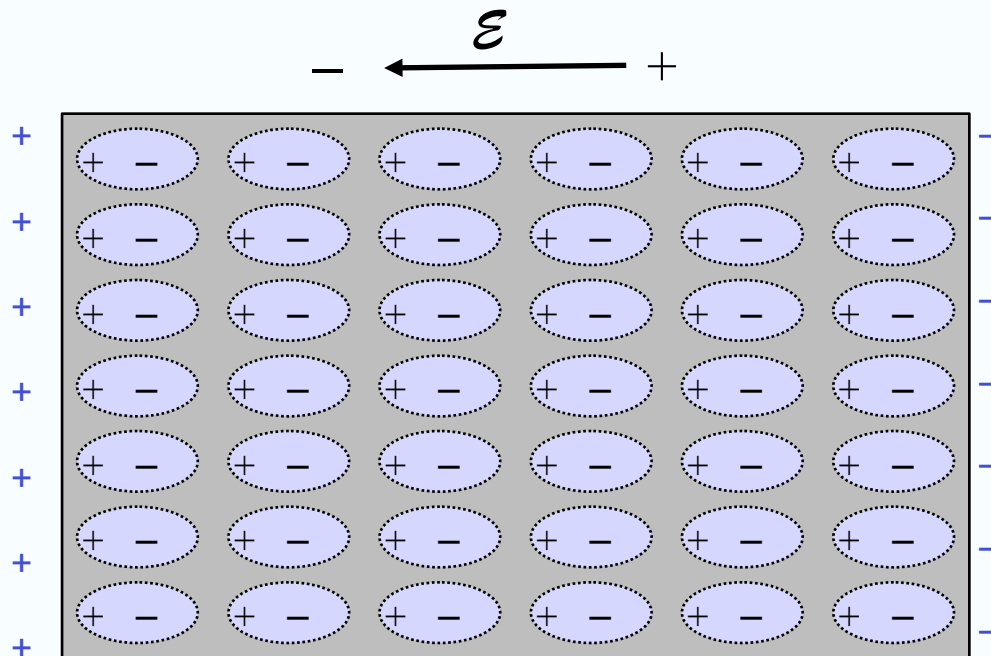
# Dipole moment

- ◆ Induced dipole moment of a single atom by the external electric field

$$\mathcal{P}_a = qd$$

- ◆ Polarisation of a dielectric crystal (dipole moment per volume)

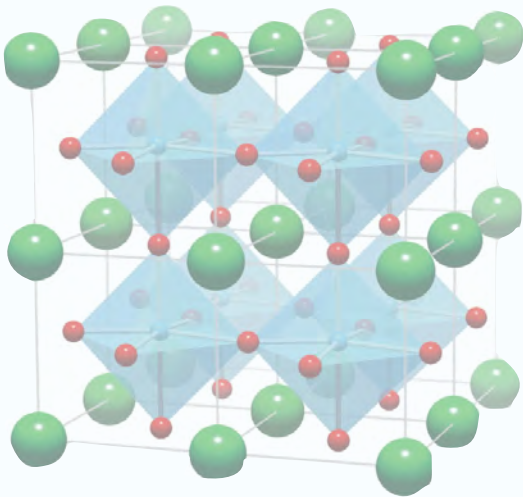
$$\mathcal{P} = Nqd$$



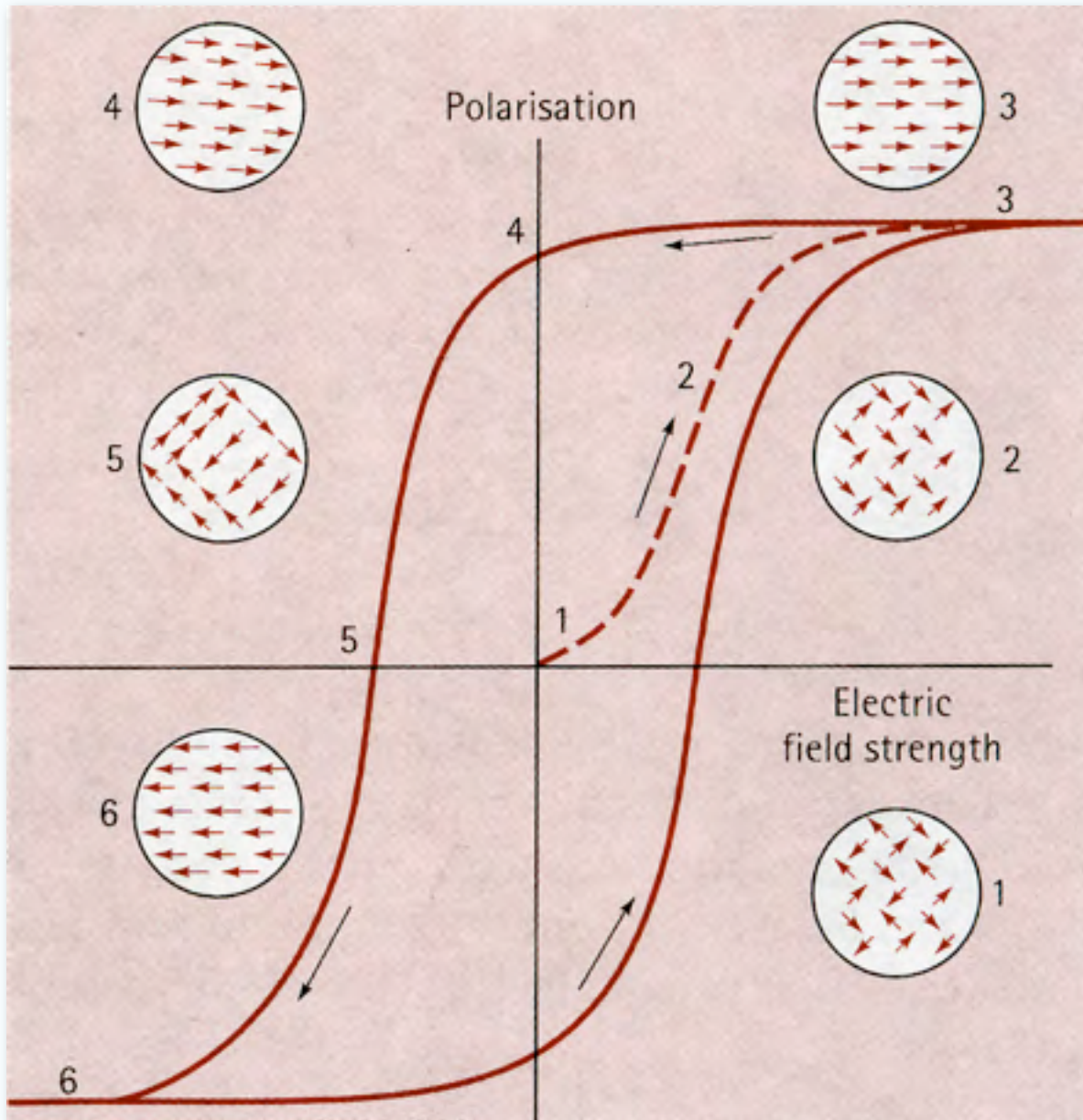
Dielectric crystal made of atomic dipoles. The result of the external field is a surface charge on both terminals.

# Ferroelectrics

- ◆ Materials with a finite polarization, even without an external field
- ◆ Name misleading: no iron; properties resemble ferromagnetic solids
- ◆ Hysteresis in the polarization



# Arrangement of dipoles

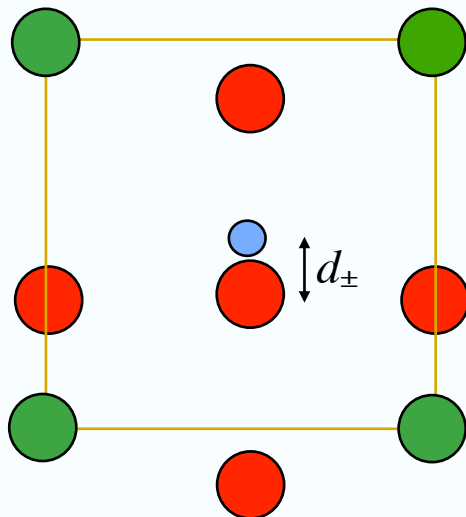
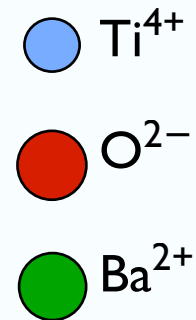
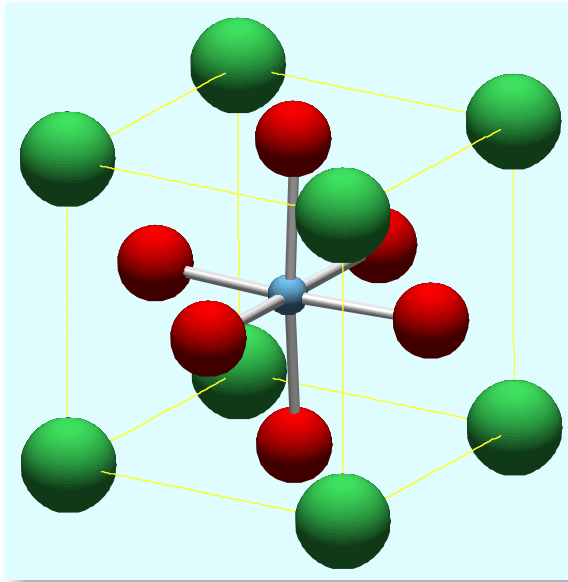


Ferroelectric hystereses with the arrangement of the dipoles

[Askeland 1996]

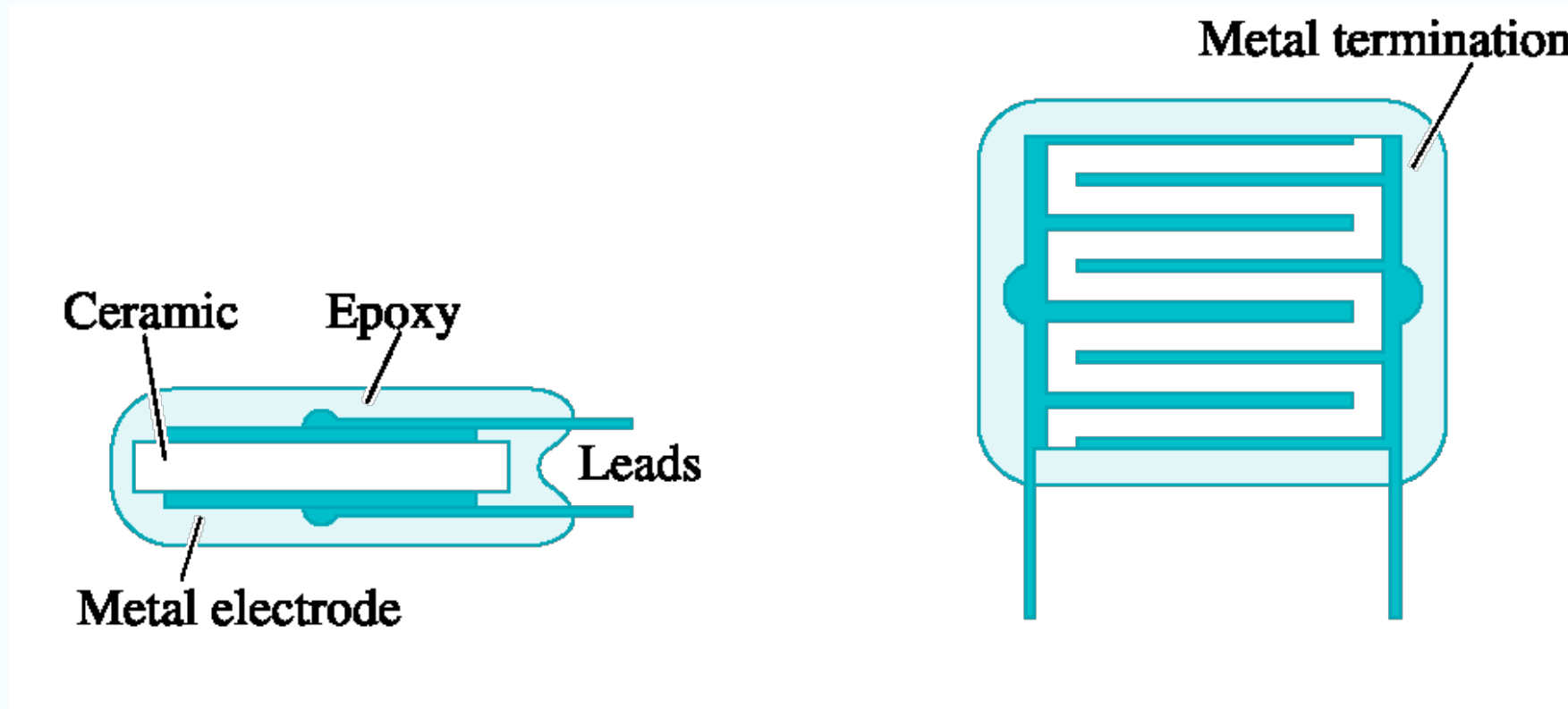


# Structure of barium titanate



Compared to the ideal cubic arrangement, the positive and negative ions are shifted by a distance of  $d_{\pm} \approx 0.01$  nm.

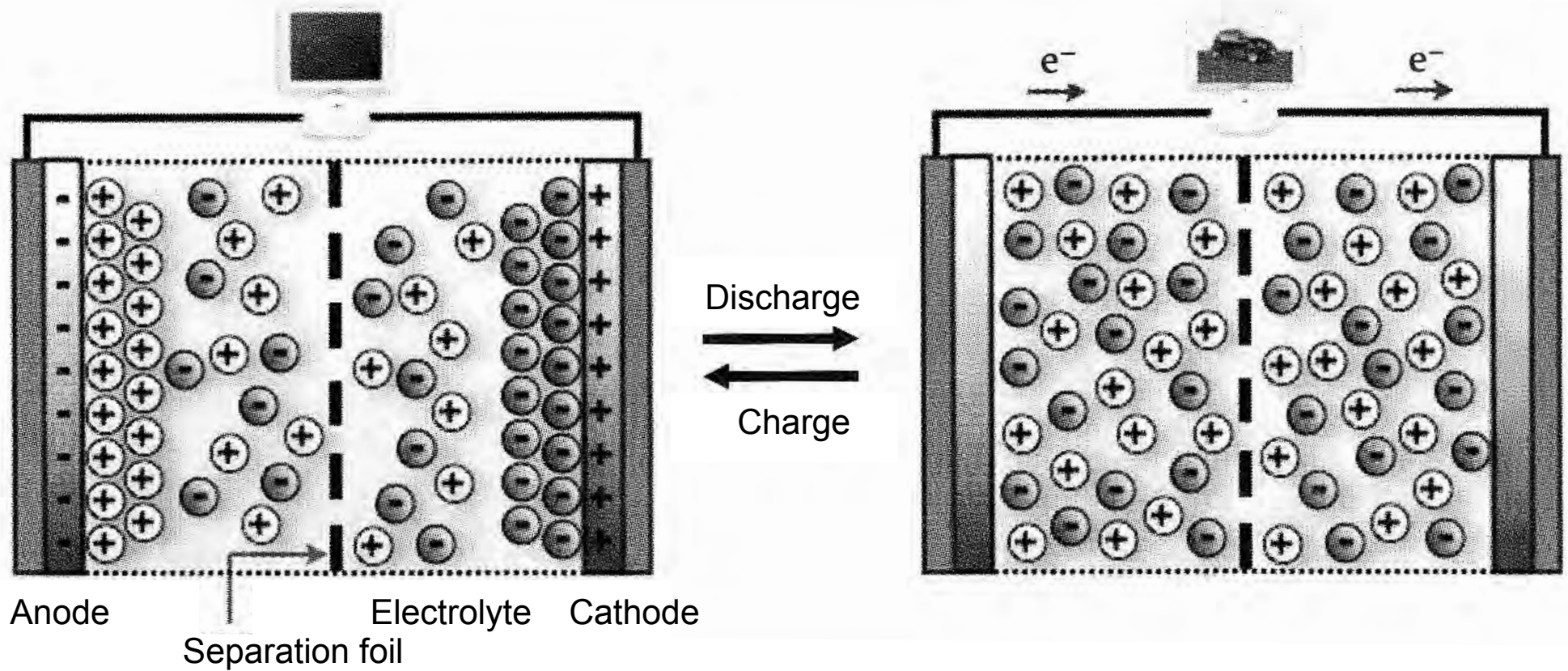
# Design of a ceramic capacitor



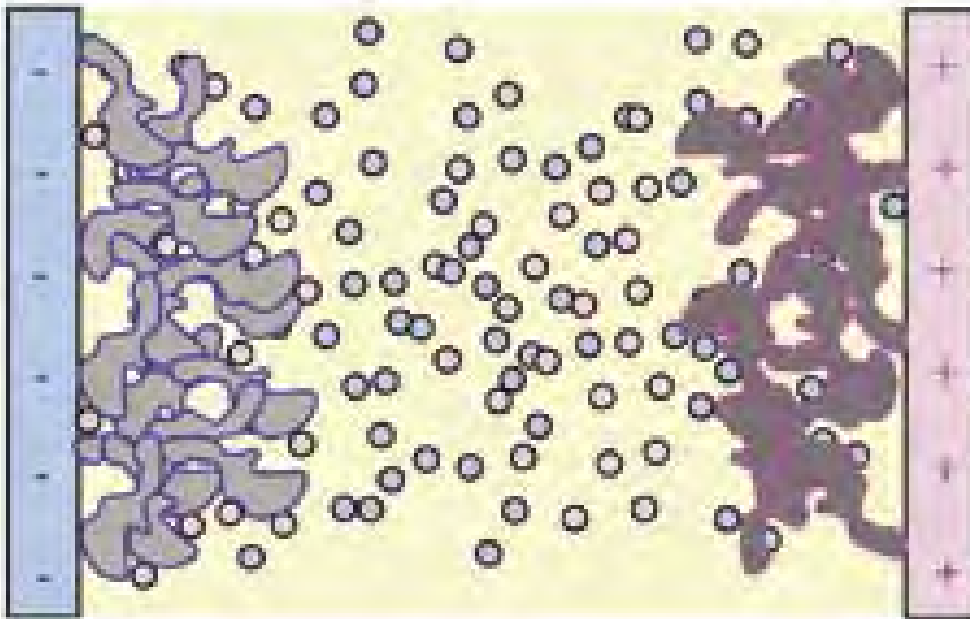
Examples of ceramic capacitors. Single-layer ceramic capacitor (plate capacitor) and multilayer capacitor (stacked ceramic layers).

[Askeland 1996]

# Double-layer capacitor



# Capacity



$$C = \frac{\epsilon_r \epsilon_0 A}{d} \quad \frac{C}{A} = \frac{\epsilon_r \epsilon_0}{p \ln \frac{p}{a_0}}$$

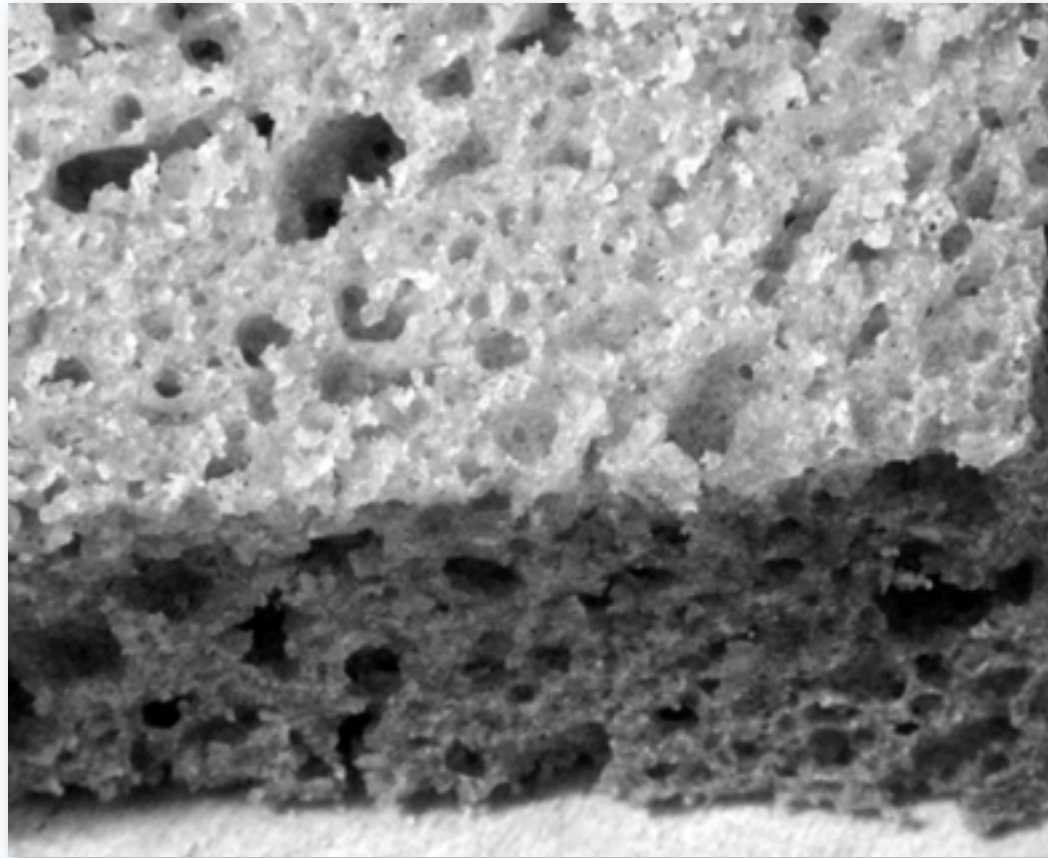
( $p$  pore radius,  $a_0$  effective ion size)

$$E = \frac{1}{2} C U^2$$

Charged double-layer capacitor with two double layers in series (i. e. the interfaces electrode–charged layer and charged layer–electrolyte) with a large specific surface.

[Scherson, Palencár 2006]

# Capacity



10  $\mu\text{m}$

Graphite particles with a large specific surface  
[Takamura *et al* 2007]

# Commercially available standard capacitors

## Ceramic capacitors

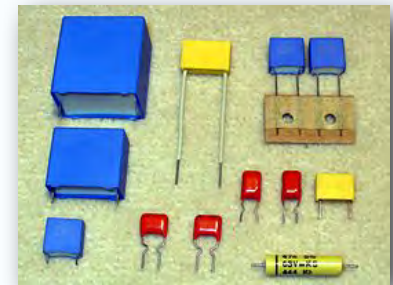
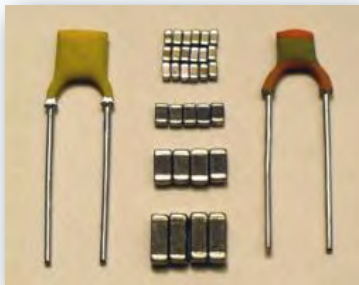
based e. g. on barium titanate

- + high permittivity
- + thermal stability
- + allow high frequencies
- brittle

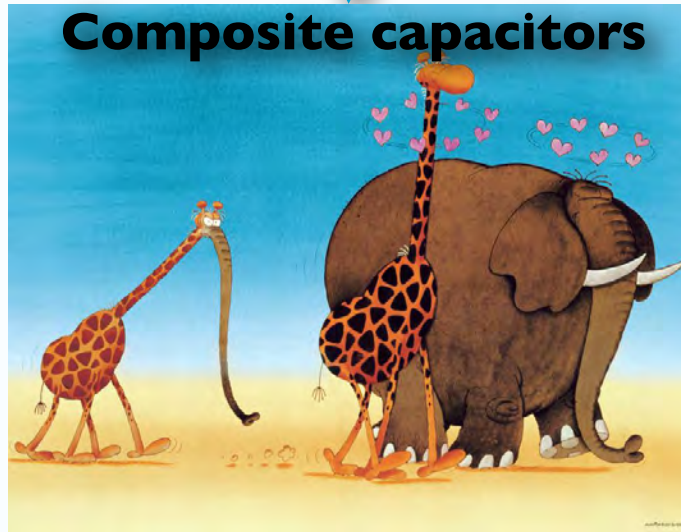
## Thin-film polymer capacitors

e. g. PET, PP

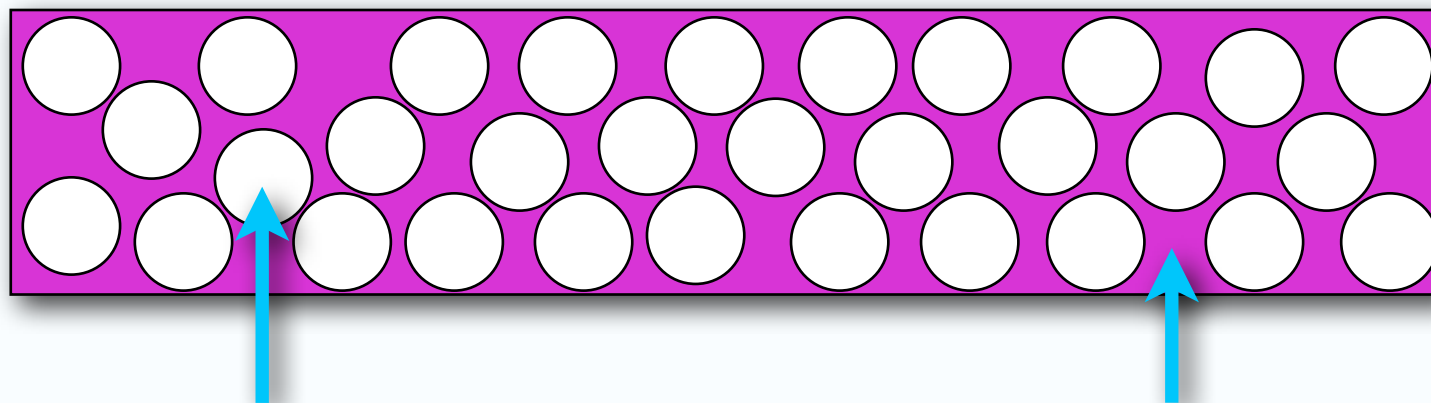
- + high voltage
- + low conductivity
- + simple shapes
- low permittivity



## Composite capacitors

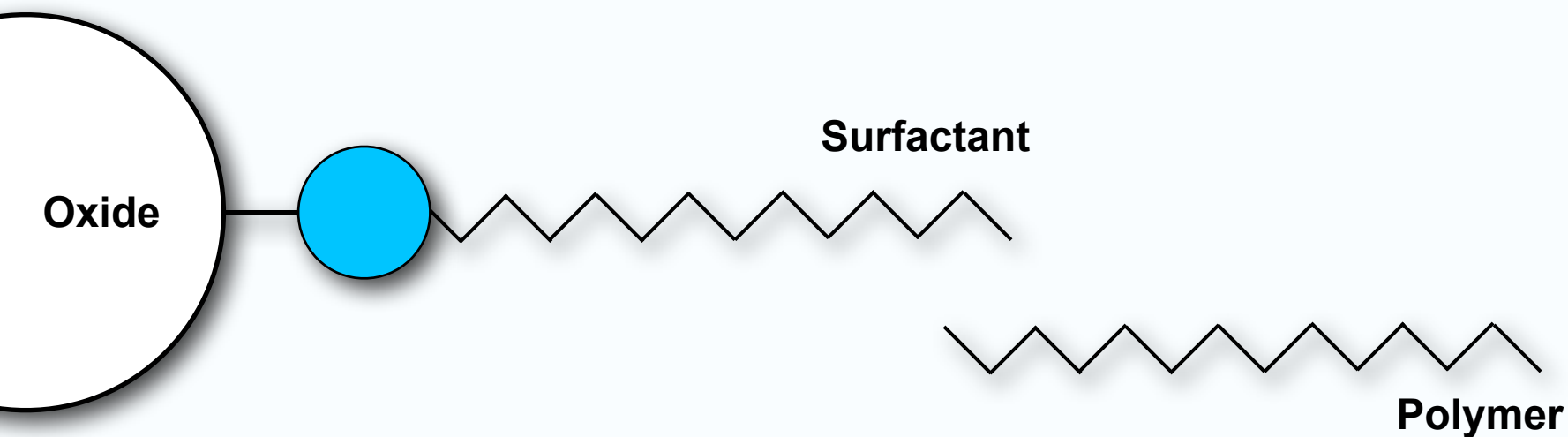


# Composite dielectrics



**Oxide particle**  
polar, hydrophilic

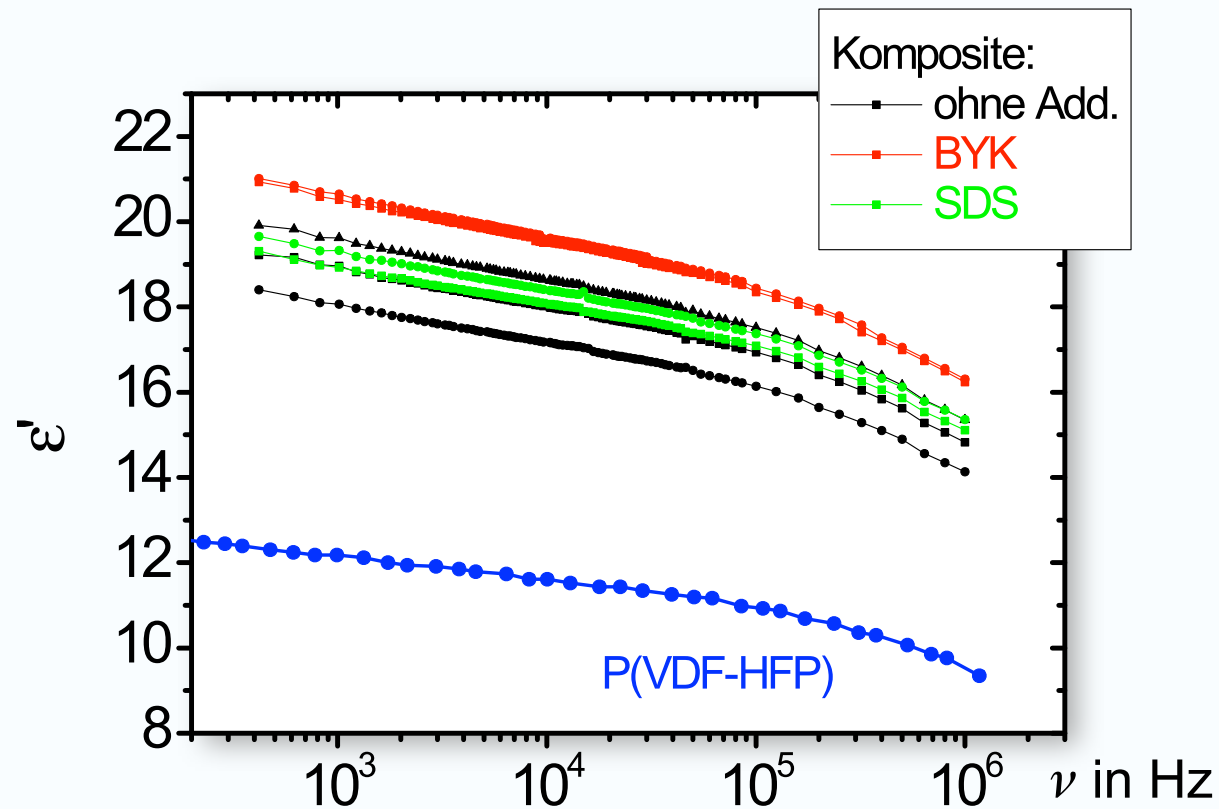
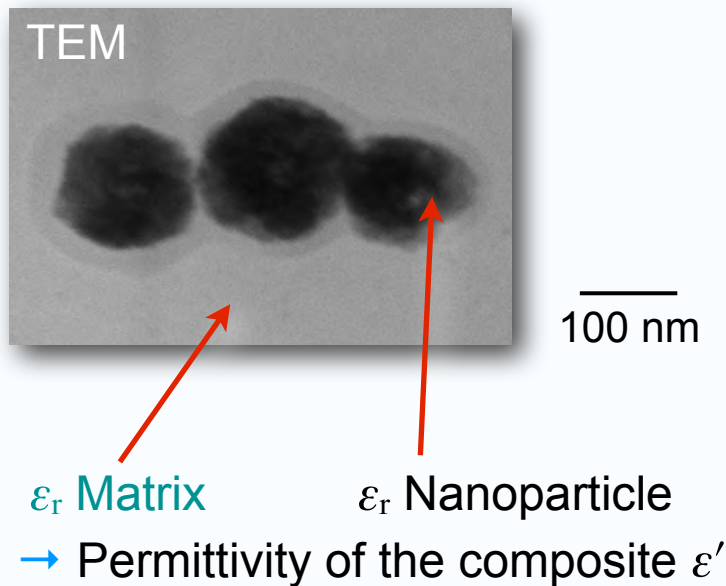
**Polymer matrix**  
nonpolar, lipophilic



# Mixing rules

## Simple models

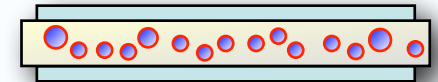
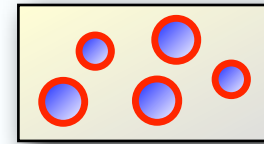
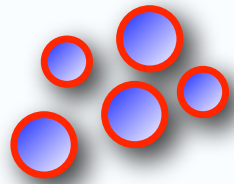
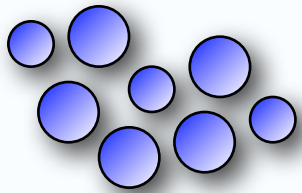
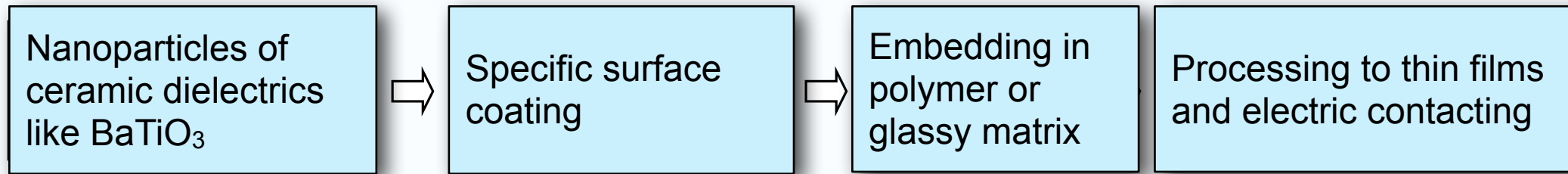
- ◆ Serial or parallel connections
- ◆ Isotropic statistic distribution of spherical particles in a homogeneous matrix



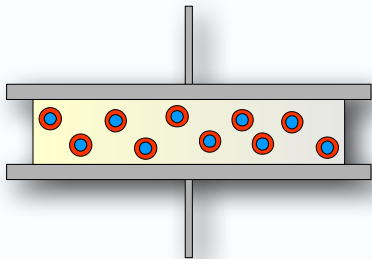
Permittivity  $\epsilon'$  as a function of the frequency  $\nu$  for different 0–3 composites



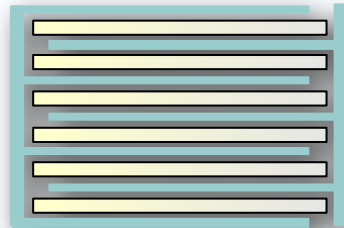
# Composite capacitors



Single capacitor



Multilayer capacitor



Assembly



Module

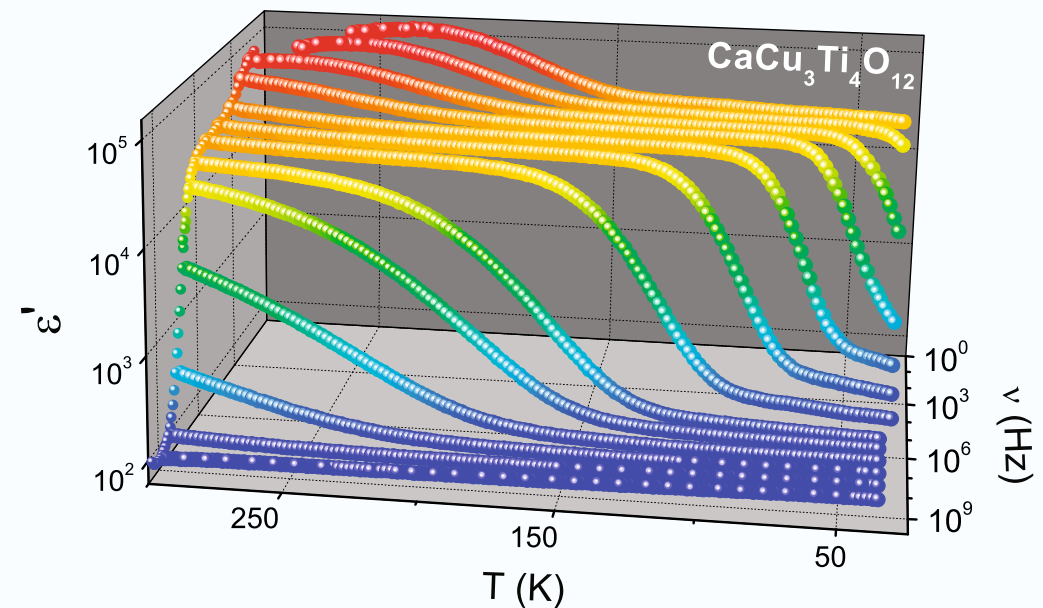
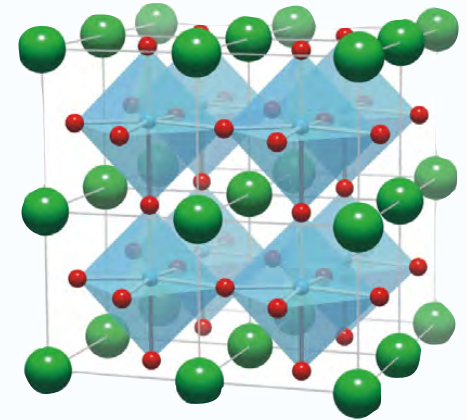


# Advantages of composite supercapacitors

- ◆ Robust, negligible aging, high lifetime
- ◆ High charging voltages
- ◆ Thermal stability (operation temperatures  $> 60\text{ }^{\circ}\text{C}$  possible)
- ◆ No cooling
- ◆ High charging or discharging rates
- ◆ High efficiency
- ◆ Modular structure
- ◆ Environmentally friendly
- ◆ Reasonable energy and power density

# Ceramic particles

- ◆  $\text{BaTiO}_3$ 
  - ❖ Ferroelectric,  $\epsilon_r > 2\,000$
  - ❖ Phase transitions
- ◆  $\text{CaCu}_3\text{Ti}_4\text{O}_{12}$ 
  - ❖ Non ferroelectric
  - ❖ Giant  $\epsilon_r > 100\,000$
- ◆ Different synthesis routes
  - ❖ Oxide mixing, Pecchini, Oxalate, Sol-Gel
  - ❖ Particle size 50...100 nm



Permittivity  $\epsilon'$  of single crystal CCTO as a function of the temperature  $T$  and the frequency  $\nu$

[Lunkenheimer *et al* 2010]

# Matrix and shell components

## ◆ Polymer films

- ❖ PVDF
- ❖ P(VDF-HFP)
- ❖ Poly(bisphenol A-carbonate)

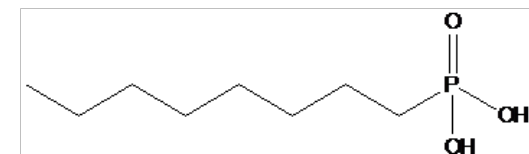
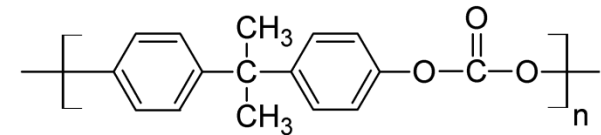
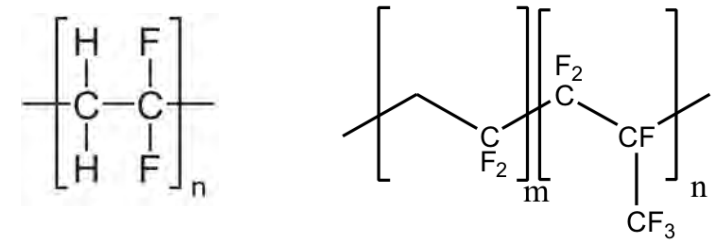
## ◆ Glasses

## ◆ Preparation methods

- ❖ Sintering, spin coating, spray deposition

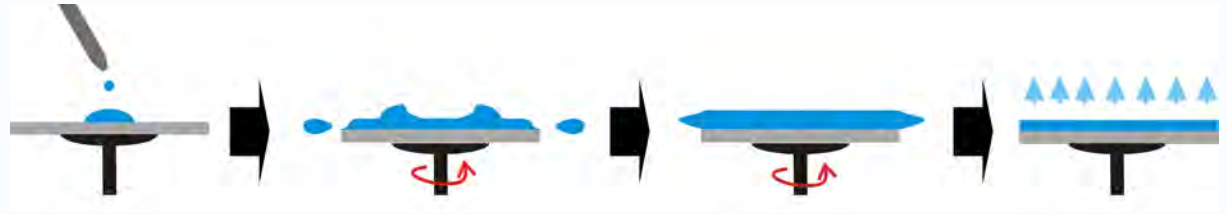
## ◆ Surface coating

- ❖ Passivation of the surface, block aggregation/percolation, minimum of leakage current, high breakdown voltage
- ❖ Phosphonic acids; E-glass

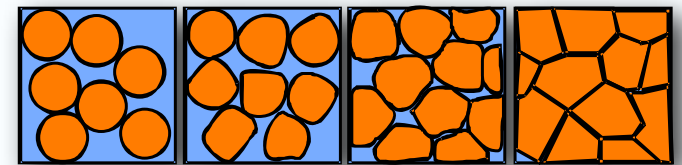


# Thin film preparation

**homogeneous, reproducible, scalable, cheap**



- ◆ Single films, lab stage
  - ❖ Spin coating
    - Established for homogeneous solutions
    - More difficult for composites
    - Thickness profile may become inhomogeneous
    - Problems with rectangular substrates, geometry effects
  - ❖ Molding, pressing sintering
- ◆ Large areas with linear coating, spray deposition
- ◆ Transition to multilayers



# Energy density of a capacitor

- ◆ Max. voltage given by the break-down voltage

$$U_b = \mathcal{E}_b d$$

- ◆ Storage density

$$w_{Sp} = \frac{E}{V} = \frac{\mathcal{C}}{Ad} \frac{U_b^2}{2} = \epsilon_0 \epsilon_r \frac{\mathcal{E}_b^2}{2}$$

- ◆ Typical no. for a polymer dielectric:  $\approx 0,3 \text{ kWh m}^{-3}$
- ◆ Storage efficiency of capacitors  $\rightarrow 1$



# Application of supercapacitors

- ◆ Control of the pitch angle of the rotors in wind turbines – big variation in  $T$ , independent of the grid, no maintenance
- ◆ Start of microturbines or fuel cells working as UPS requires usually some 100 kJ electrical energy within ca. 10 ... 20 s
- ◆ Energy storage for photovoltaics; capacitors can supply periodic power with higher currents as coming directly from the solar modules
- ◆ Recuperation of brake energy in cars



# Recuperation

- ◆ Brake which regains the electrical energy from the motor acting as a generator during braking (grid, energy storage)
- ◆ Since the 1920ies in the Swiss *Krokodil*
- ◆ Hybrid cars: electrical energy into the battery, storage capacitors or fly wheel



SBB Ce 6/8<sup>II</sup> electrical locomotive  
"Krokodil", working in the  
Gotthard railway until the 1980ies

recuperare (*lat.*) = regain



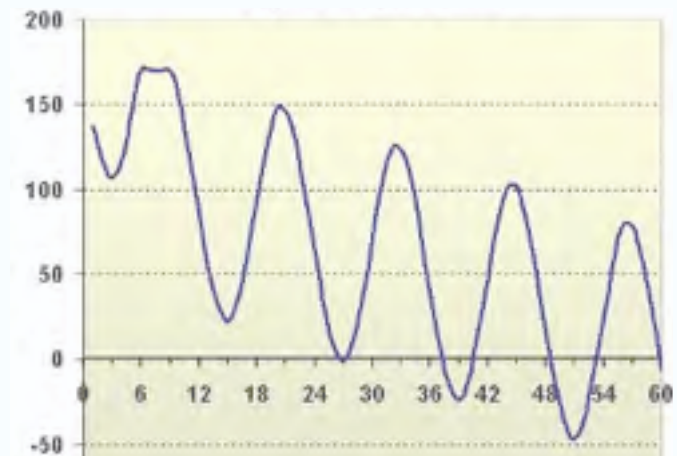
# Fuel saving in the car

- ◆ Regain of kinetic energy when idling or braking and feeding into the battery
- ◆ During acceleration all energy consumers which are not necessary are separated from the power train
- ◆ strong generator + electronic regulation: „dynamo“ not working permanently
- ◆ Commercials: *Efficiency dynamics, brake energy regain*



# Benefits of capacitors for energy storage

- ◆ Maintenance free, relative low weight
- ◆ Resistent to temperature variations
- ◆ Long lifetime
- ◆ More than 500 000 charge–discharge cycles
- ◆ No destruction by deep discharge



# Conclusions

- ◆ Present energy density of capacitors low, but high power density
- ◆ Possibilities of short-term storage (e. g. grid stabilization, automotive applications, sensors)
- ◆ High development potential with the overcome of materials science problems



“Did anyone call for high-power, infinitely rechargeable electrical energy storage?”

**Thanks to the Super-Kon team:**

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T. Großmann, S. Lemm, W. Münchgesang,  
C. Pientschke, K. Suckau, G. Wagner, M. Zenkner



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